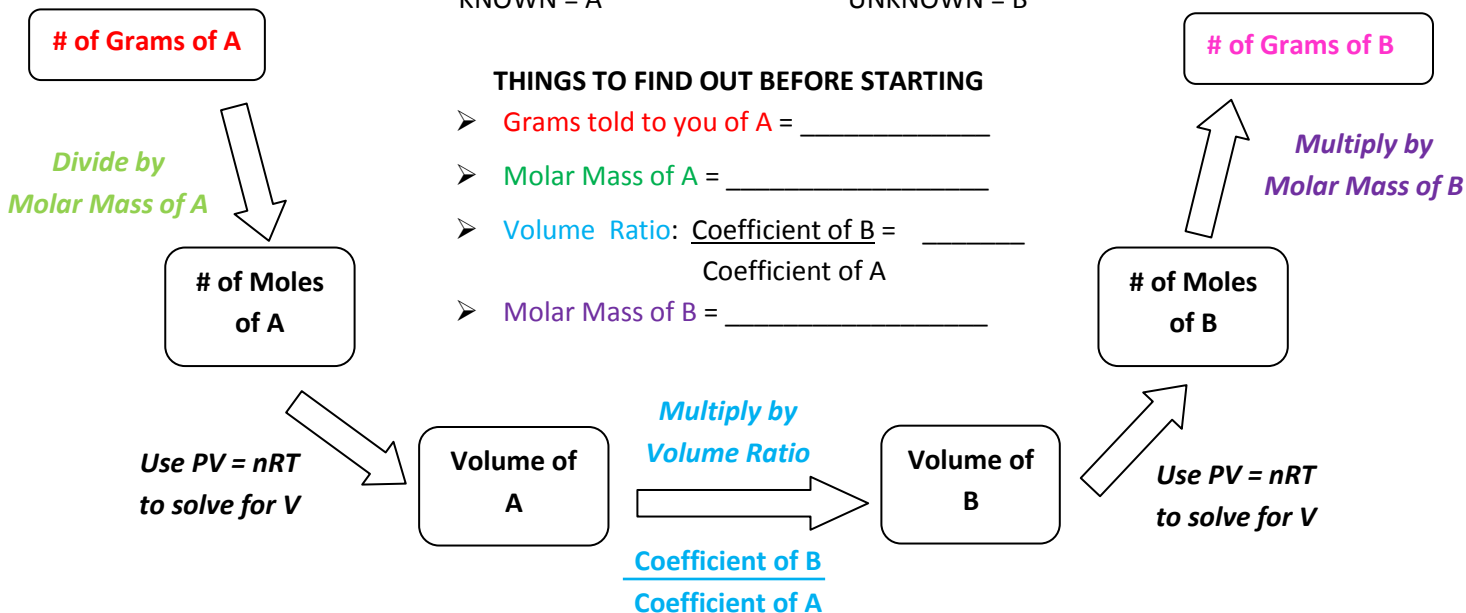


## Gas Stoichiometry Flow Chart

If you have X grams of A, how many grams of B can you make?

KNOWN = A

UNKNOWN = B



### THINGS TO FIND OUT BEFORE STARTING

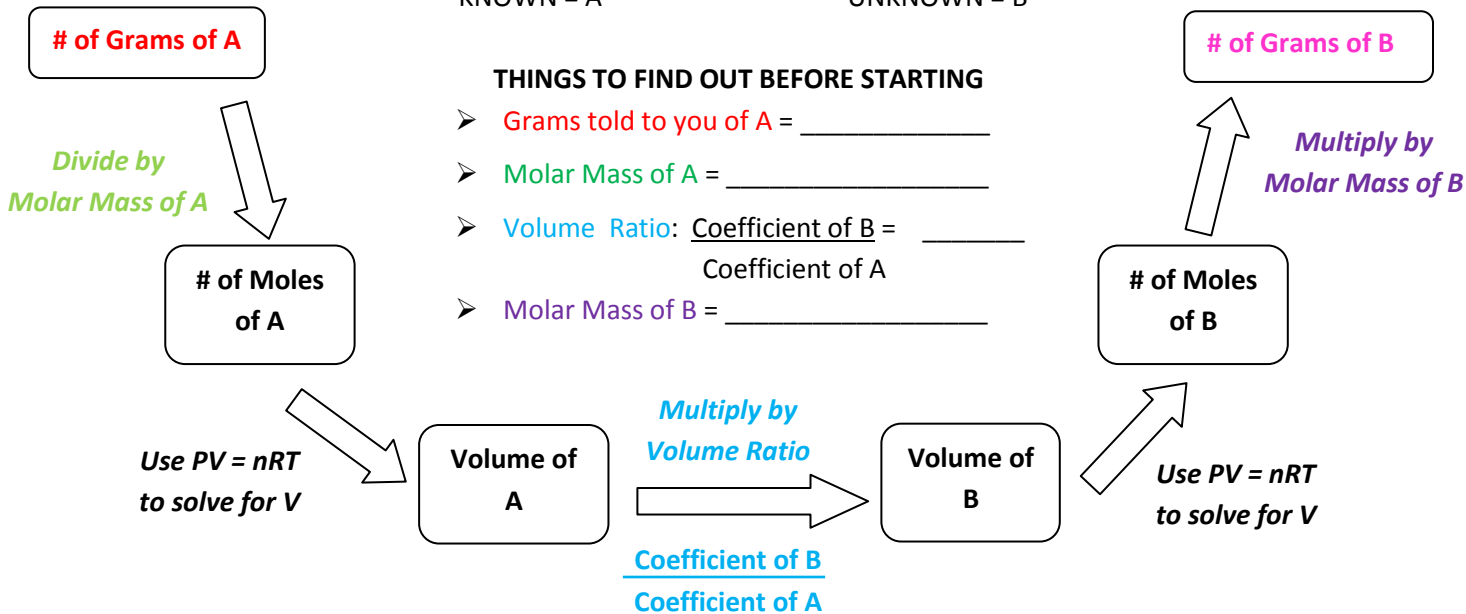
- Grams told to you of A = \_\_\_\_\_
- Molar Mass of A = \_\_\_\_\_
- Volume Ratio:  $\frac{\text{Coefficient of B}}{\text{Coefficient of A}} =$  \_\_\_\_\_
- Molar Mass of B = \_\_\_\_\_

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